## **Unit 1: Fire Engineering Science**

### Introduction

This unit focuses on fire engineering science and fire behaviour. The content of the unit has been designed to reflect the critical technical knowledge that fire professionals need in order to understand the behaviour of fire and the mechanics of firefighting equipment. This knowledge and understanding will contribute to increased safety on the incident ground.

### **Learning Outcomes**

Candidates who achieve this unit should be able to:

- Interpret data and carry out relevant calculations
- Understand the scientific principles that underpin fire behaviour and the management of fires
- Understand and apply hydraulics in firefighting contexts

### **Unit Status**

This is a mandatory unit for candidates who wish to achieve the Level 3 Diploma in Fire Science and Fire Safety.

### **Content**

### 1. Analysis and Interpretation of Data

Assessment Objective	Knowledge, Understanding and Skills
1.1 Extract and tabulate data	Expression of data in the form of:
	Graphs including histograms, bar charts, pie charts
	Tables
1.2 Obtain values from data	Identify/Calculate
	Median
	Mean
	Norm values
1.3 Extend graphs	Project values from given data (extrapolate)
	<ul> <li>Deduce values from missing data (interpolate)</li> </ul>

## 2. Mechanics

Assessment Objective	Knowledge, Understanding and Skills
2.1 Define and apply the SI system of units in terms of basic and derived units	Recognise and use SI units for calculation and expressing values
2.2 Describe and carry out simple calculations involving equations of motion	Describe and apply Newton's laws of motion
2.3 Understand and solve calculations involving vectors	<ul><li>Use vector quantities to find resultant values</li><li>Apply vector methods to force and motion problems</li></ul>
2.4 Calculate moments around a fulcrum including the use of levers and parallel force	<ul> <li>Definition of "fulcrum"</li> <li>Definition of "moment"</li> <li>Methods of calculation</li> </ul>
2.5 Carry out calculations involving centres of gravity and centres of buoyancy	<ul> <li>Definition of "centres of gravity"</li> <li>Definition of "centres of buoyancy"</li> <li>Methods of calculation</li> </ul>
2.6 Carry out calculations involving stress and strain	<ul><li>Definition of stress strain</li><li>Understand and apply Hooke's Law of elasticity</li></ul>
2.7 Carry out calculations involving work, power and efficiency	Understand and apply the definitions of:  O Work O Power O Efficiency O Force O Momentum O Mass O Weight
2.8 Describe and calculate friction force between two surfaces in contact	<ul> <li>Definition of "friction"</li> <li>Methods of calculation</li> </ul>

# 3. Hydraulics

Assessment Objective	Knowledge, Understanding and Skills
3.1 Define and solve problems involving density, specific gravity and pressures in fluids	<ul> <li>Define the terms density, specific gravity and pressures in fluids</li> <li>Understand and apply the definitions of velocity, acceleration and energy</li> <li>Demonstrate the relationship between the terms</li> <li>Methods of calculation</li> </ul>
3.2 Understand and apply the principle of atmospheric pressure in pumping systems	<ul> <li>As an aid to flow</li> <li>As a means of measuring flow</li> <li>Definition of "atmospheric pressure" and methods of measuring it</li> </ul>
3.3 Understand and apply the laws of friction to calculate energy losses in piped water supplies	<ul> <li>Laws of friction</li> <li>Methods of calculation</li> <li>Operation of piped water supplies</li> </ul>
3.4 Understand and explain the	Definition of water power

operation of pumps and carry out basic calculations	<ul> <li>Definition of brake power</li> <li>Definition of efficiency</li> </ul>
	Types of calculation required and methods of calculation
3.5 Explain the relationship between velocity and discharge of water through hose of differing diameters	<ul> <li>Methods of calculating velocity, flow and quantity of water in hose and pipelines of differing diameters</li> </ul>
3.6 Explain the purpose and principles of design of siphons, branches and nozzles	<ul> <li>Purpose of nozzles and siphons</li> <li>Design and operating principles of nozzles</li> <li>Design and operating principles of siphons</li> <li>Methods of calculating discharge from a nozzle</li> </ul>
3.7 Calculate the theoretical and the effective height of a jet	<ul> <li>Methods of calculation</li> <li>Significance of difference in specific gravities between different liquids</li> </ul>

# 4. Electricity

Assessment Objective	Knowledge, Understanding and Skills
4.1 Understand the theory of	Define the terms "electron" and "current"
electrical current flowing in a	Describe electric current as a flow of electrons
circuit and apply this	Describe how electrical energy is generated and distributed
	Describe the characteristics of alternating and direct current
	<ul> <li>Define the following: Volts, Amperes, Ohms, Watts, Joules</li> </ul>
	Methods of calculating current, power, voltage, energy and resistance
4.2 Describe the operation and	Principles of operation of a transformer
characteristics of a step-up and	Methods of calculation
step-down transformer	
4.3 Explain and use Ohm's Law	Principles of Ohm's Law
	Use Ohm's Law to solve problems
	Calculate the relationship between resistance, current
	and voltage in simple parallel and series circuits
4.4 Explain the magnetic and	Show how these phenomena are applied in:
chemical effects of electrical	a) Electric motors
currents	b) Primary and secondary electric cells
4.5 Describe the function and	Types to include:
method of operation of circuit	Residual Current Device (RCD)
protective devices	Miniature circuit breakers (MCB)
	• Fuses
4.6 Define and solve problems	Temperature
involving resistance variation	Resistivity
	Methods of calculation
4.7 Apply the concept of power to	Power formula
electrical circuits	Methods of calculation

## 5. Heat

Assessment Objective	Knowledge, Understanding and Skills
5.1 Explain the principles of heat transfer	<ul> <li>Methods of heat transfer</li> <li>Definition of:         <ul> <li>Absolute zero</li> <li>Specific heat capacity</li> <li>Latent heat</li> <li>Thermal capacity of a body</li> </ul> </li> <li>Methods of calculation</li> </ul>
5.2 Calculate linear, superficial and volumetric expansion using the relevant coefficients	<ul> <li>Definition of linear, superficial and volumetric expansion</li> <li>Understand and use coefficients</li> </ul>
5.3 Apply the Gas Laws to calculations involving changing conditions of heat	Define and use Gas Laws:  Boyle's Law Charles's Law Law of Pressures (also known as Gay-Lusacc's Law) Combined Gas Law

# 6. Radioactivity

Assessment Objective	Knowledge, Understanding and Skills
6.1 Describe the principle of	Explain the terms:
radioactivity	<ul> <li>Radioactivity</li> </ul>
	<ul> <li>Radiation</li> </ul>
	Define the terms decay and half life
6.2 Describe the biological effects of radiation and precautions to be	Explain the construction and properties of alpha and beta particles and gamma radiation and compare their
adopted for safety from the	penetrating powers
effects of radiation	Effects on cells
	Methods of contamination
	Principles of protection from ionising radiation

# 7. Chemistry

Assessment Objective	Knowledge, Understanding and Skills
7.1 Describe the construction of an atom and show how the electron shell configuration has an effect on reactivity	<ul> <li>Definition of "reactivity"</li> <li>Components of an atom – protons, neutrons, electrons, shell</li> </ul>
7.2 Demonstrate an understanding of the classifications of the chemical elements and the main hazards associated with each grouping	<ul> <li>Structure of the Periodic Table of Elements</li> <li>Classification of elements into Metals and Non-metals</li> <li>Properties of elements:         <ul> <li>Reaction to heat</li> <li>Reaction to electricity</li> <li>Physical form in different situations</li> <li>Reaction with oxygen</li> <li>Reactivity</li> </ul> </li> </ul>

	Explain the concept of valency and the relevance of the periodic classification of the elements
7.3 Demonstrate an understanding of basic chemical equations	<ul> <li>Balance simple chemical equations and define stoichiometric conditions</li> <li>Calculate relative molecular masses and vapour densities from given relative atomic masses</li> <li>Use "balanced" chemical reactions for the calculation of the masses and the volumes, of reactants in chemical reactions</li> </ul>
7.4 Describe the main properties, reactions and hazards of specified elements, compounds and groups	<ul> <li>Acids (inorganic and organic)</li> <li>Bases and Alkalis</li> <li>Ammonia</li> <li>Calcium Oxide</li> <li>Ammonium hydroxide</li> <li>Carbon monoxide</li> <li>Carbon dioxide</li> <li>Chlorine</li> <li>Hydrogen</li> <li>Oxygen</li> <li>Sodium</li> <li>Sulphur</li> <li>Phosgene</li> </ul>
7.5 Describe the properties of hydrocarbons	<ul> <li>Composition of hydrocarbons - hydrogen and carbon</li> <li>Structure and main properties of the first four members of the alkane family         <ul> <li>Methane, ethane, propane and butane</li> <li>Properties – density, boiling point and melting point</li> </ul> </li> <li>Structure of simple unsaturated hydrocarbons (alkenes and alkynes)</li> </ul>
7.6 Define the terms flashpoint, fire point and spontaneous ignition temperatures	<ul> <li>Definition of flashpoint</li> <li>Definition of fire point</li> <li>Definition of spontaneous ignition temperatures</li> </ul>
7.7 Apply the principles of chemistry to the extinction of fire	<ul> <li>Principles involved in the extinction of fire by:         <ul> <li>Smothering</li> <li>Cooling</li> <li>Oxygen starvation</li> </ul> </li> <li>Understand that a combustion is a type of chemical reaction</li> <li>Principle of the fire tetrahedron and inhibition of combustion chains involving a free radical mechanism</li> <li>Components of the tetrahedron: fuel, heat, oxygen, chemical chain reaction</li> <li>Principle of free radicals</li> </ul>
7.8 Describe the properties of oxidising agents	<ul> <li>Oxygen</li> <li>Halogens</li> <li>Inorganic and organic oxidising agents</li> </ul>
7.9 Describe the properties of	<ul><li>Peroxide</li><li>Define monomer and polymer</li></ul>

polymers	•	Explain the polymerisation process
	•	Thermosetting and thermoplastic materials
	•	Fire hazards

# 8. Principles of Heat and Combustion Sensitive Detection Devices

Assessment Objective	Knowledge, Understanding and Skills
8.1 Describe in detail the operating	Types of device:
principles of heat and	<ul> <li>Ionisation detectors</li> </ul>
combustion sensitive detection	<ul> <li>Optical detectors</li> </ul>
devices	<ul> <li>Heat detectors</li> </ul>
	<ul> <li>Combustion detectors</li> </ul>
	<ul> <li>Radiation detectors</li> </ul>
	<ul> <li>Flame detectors</li> </ul>
	Use and effectiveness of the detectors listed above
	according to the risk to be covered and their reliability
8.2 Describe the principles of	Explain the use of thermocouples
thermocouples and thermistors	Explain the use of thermistors